

# Observation of Enzyme Activity



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# Outline



- What are enzymes
- Different examples of enzymes
- Main features of enzymes
- Experimental Work: Enzyme Activity
- Materials
- Procedure
- Expected Results

# ■ Objectives

By the end of this lab, students should be able to:

Define enzymes and explain their biological importance

Describe the main characteristics of enzymes

Observe and explain enzyme activity in a laboratory experiment

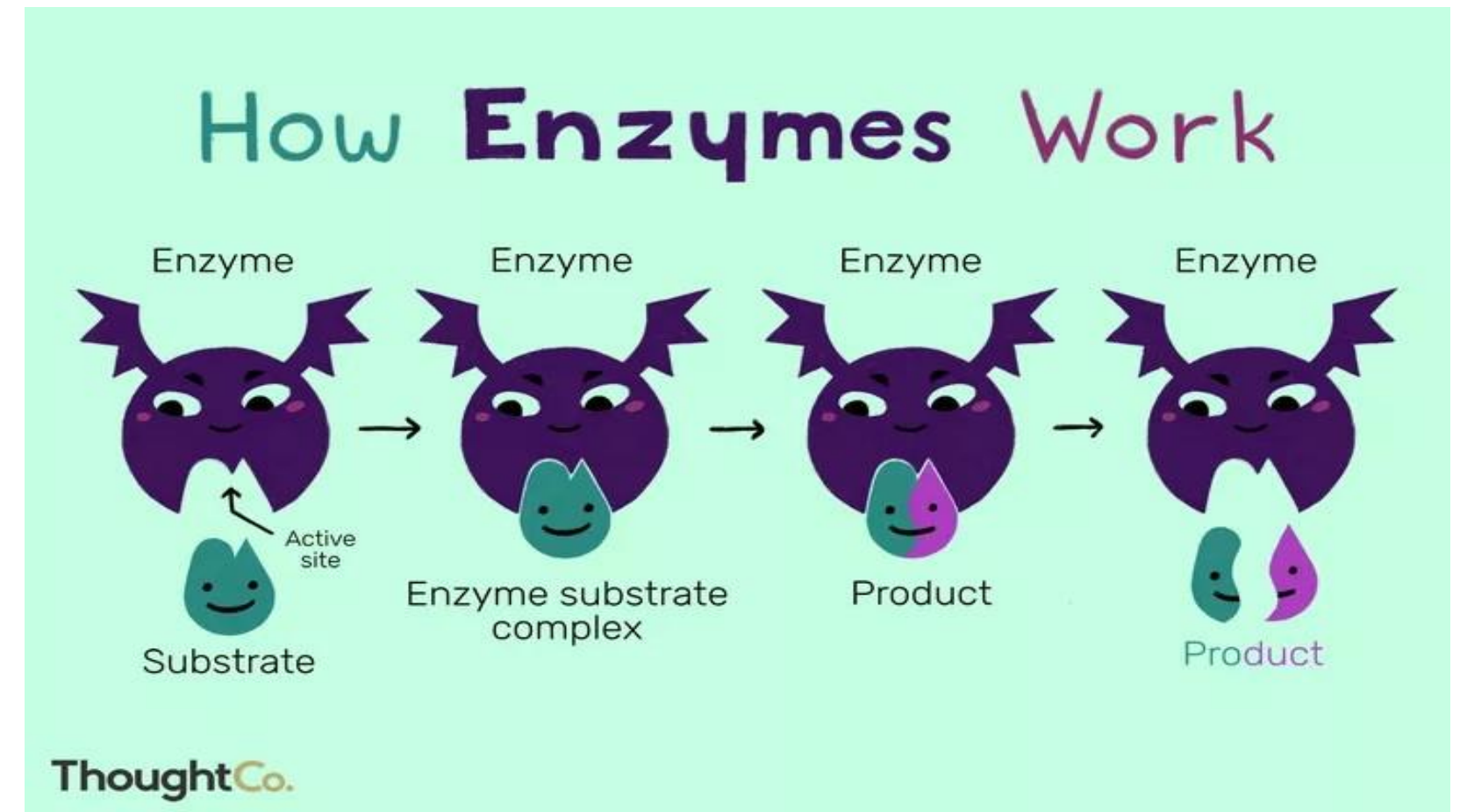
# ❖ What are enzymes?

- Are biological catalysts, usually proteins, that speed up chemical reactions without being consumed.
- Enzymes:
- **Do not change the final products**
- **Do not change reaction equilibrium**
- **Only reduce activation energy**

# ❖ How Enzymes Work?



- They work by binding to a specific substrate at the active site to form an enzyme–substrate complex.



# ❖ Examples of Digestive Enzymes:



- Break down food into absorbable molecules.

Enzyme	Reaction	Site
Amylase	Starch → maltose	Saliva, pancreas
Pepsin	Proteins → peptides	Stomach
Trypsin	Proteins → peptides	Small intestine
Lipase	Triglycerides → fatty acids + glycerol	Pancreas
Lactase	Lactose → glucose + galactose	Intestine

# ❖ Main features:



## 1. Highly specific

Each enzyme acts on only one specific substrate or a very small group of related substrates.

## 2. Reusable

Enzymes are not used up during the reaction.

One enzyme molecule can catalyze the same reaction many times.

# ❖ **Main features:**



## **3. Work best under optimum conditions**

Each enzyme has specific conditions where it works most efficiently.

## **4. Sensitive to environmental changes**

Small changes in temperature, pH, or chemicals can affect enzyme activity.

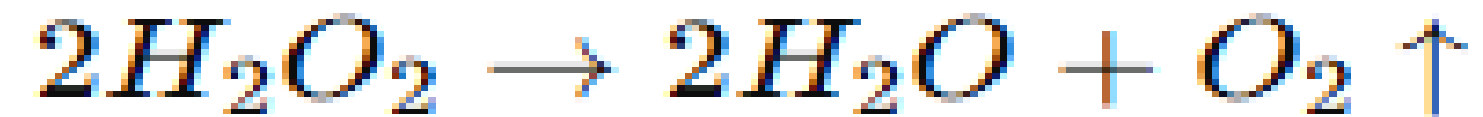


# ❖ Laboratory Experimental Work: Enzyme Activity



- Demonstration of enzyme activity using catalase.
- **Aim:** To demonstrate enzyme activity and investigate how enzymes catalyze biochemical reactions using catalase as a model enzyme.
- **Principle:** Enzymes are biological catalysts that speed up chemical reactions without being consumed.

- Catalase is an enzyme widely found in living cells.
- It protects cells from oxidative damage by breaking down hydrogen peroxide into harmless substances



- The release of oxygen gas (seen as bubbles) is direct evidence of enzyme activity.
- The rate and amount of bubbling reflect the level of catalase activity.

## ❖ **Materials Required:**

- Fresh potato, yeast, or liver extract (source of catalase)
- 3% hydrogen peroxide solution
- Test tubes
- Test tube rack
- Measuring cylinder or graduated pipette
- Dropper
- Mortar and pestle/blender (for extract preparation)
- Filter paper or gauze
- Beaker
- Stopwatch

# ❖ Procedure:

## 1. Preparation of enzyme extract:

1. Cut fresh potato/liver into small pieces.
2. Grind with a small amount of distilled water.



## Enzyme activity test :

### 1. Label two test tubes:

- Test
- Control

### 2. Add 2 mL enzyme extract to the Test tube.

### 3. Add 2 mL distilled water to the Control tube.

### 4. Add 2 mL hydrogen peroxide to both tubes.

### 5. Observe immediately.

### 6. Record the intensity of bubble formation.



## ❑ Observation:



Tube	Contents	Observation
Test	Enzyme + H <sub>2</sub> O <sub>2</sub>	Vigorous bubbling
Control	Water + H <sub>2</sub> O <sub>2</sub>	Little or no bubbling



## ❖ Expected Results:



- The test tube will show rapid bubble formation due to oxygen release.
  - The control tube will show little or no reaction.
  - This confirms the presence and activity of catalase.
- Conclusion: Catalase is an active biological enzyme that accelerates the breakdown of hydrogen peroxide into water and oxygen.
- The visible evolution of gas confirms enzyme-catalyzed reactions.

# References



- Reece, J. B., et al. (2014). Campbell Biology, 10th Ed. Pearson.
- Alberts, B. et al. (2015). Molecular Biology of the Cell, 6th Ed. Garland Science.
- Svitil, K. (2006). "Crenation and Osmosis in Cheek Cells". The American Biology Teacher, 68(6), 364–367.
- Oparka, K. J. (1994). Plasmolysis: New insights into an old process. New Phytologist, 126(4), 571–591. <https://doi.org/10.1111/j.1469-8137.1994.tb02958.x>





**Thanks**